



# Grunnlag for fastsettelse av grenseverdi

Grunnlagsdokument for  
nitrogendioksid (NO<sub>2</sub>)

Kommisjonsdirektiv 2017/164/EU

Tittel: Grunnlag for fastsettelse av grenseverdi.  
Grunnlagsdokument for nitrogendioksid (NO<sub>2</sub>).

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Denne rapporten omhandler det toksikologiske grunnlaget og vurderinger, samt tekniske og økonomiske hensyn for fastsettelse av grenseverdi for nitrogendioksid (NO<sub>2</sub>).



# Innhold

Innhold	3
Forord	4
Innledning	5
1. Stoffets identitet	5
2. Fysikalske og kjemiske data	5
2.1. Forekomst og bruk	6
3. Grenseverdier	6
3.1 Nåværende grenseverdi	6
3.2. Grenseverdi fra EU	6
3.3 Grenseverdier fra andre land og organisasjoner	7
3.4 Stoffets klassifisering	8
3.5 Biologisk overvåking	8
4 Toksikologiske data og helseeffekter	8
4.1. Kommentarer fra TEAN	8
5. Bruk og eksponering	10
5.1. Opplysning fra Produktregistret	10
5.2. Eksponering og måledokumentasjon	10
5.2.1. EXPO- data	10
5.2.2. Prøvetakings- og analysemetode	11
6. Vurdering	11
7. Konklusjon med forslag til ny grenseverdi	13
8. Ny grenseverdi	13
9. Referanser	14
Vedlegg: SCOEL/SUM/53	15



# Forord

Grunnlagsdokumenter for fastsettelse av grenseverdier utarbeides av Arbeidstilsynet i samarbeid med Statens arbeidsmiljøinstitutt (STAMI) og partene i arbeidslivet (Næringslivets hovedorganisasjon/Norsk Industri og Landsorganisasjonen i Norge) i henhold til *Strategi for utarbeidelse og fastsettelse av grenseverdier for forurensninger i arbeidsatmosfæren*. Dette dokumentet er utarbeidet ved implementering av kommisjonsdirektiv 2017/164/EU fastsatt 31. januar 2017.

EU-rådets direktiv 98/24/EC (Vern av helse og sikkerhet til arbeidstakere mot risiko i forbindelse med kjemiske agenser på arbeidsplassen) av 7. april 1998 stiller krav om at EU- kommisjonen skal legge frem forslag til indikative grenseverdier for eksponering av visse kjemikalier som medlemslandene må innføre på nasjonalt nivå. De nasjonale grenseverdiene kan være høyere enn de som står oppført i direktivet, dersom et medlemsland mener at det er nødvendig av tekniske og/eller økonomiske hensyn, men landene bør nærme seg den indikative grenseverdien. Direktivet stiller krav om at indikative grenseverdier vedtas gjennom kommisjonsdirektiv.

I Norge ble de indikative grenseverdiene innført som veiledende administrative normer. Da nye Arbeidsmiljøforskrifter trådte i kraft 1.1.2013 ble de veiledende administrative normene forskriftsfestet i forskrift om tiltaks- og grenseverdier og fikk betegnelsen tiltaksverdier. I 2015 ble begrepet «grenseverdi» for kjemikalier presisert og begrepet «tiltaksverdi» for kjemikalier ble opphevet i forskrift om tiltaks- og grenseverdier. I vedlegg 1 til forskriften ble det innført en tydeliggjøring av anmerkningene.

Arbeidstilsynet har ansvaret for revisjonsprosessen og utarbeidelse av grunnlagsdokumenter for stoffene som blir vurdert. Det toksikologiske grunnlaget for stoffene i denne revisjonen baserer seg i hovedsak på kriteriedokumenter fra EUs vitenskapskomité for fastsettelse av grenseverdier, Scientific Committee for Occupational Exposure Limits (SCOEL). SCOEL utarbeider de vitenskapelige vurderingene som danner grunnlaget for anbefalinger til helsebaserte grenseverdier, og disse legges fram for kommisjonen.

Statens arbeidsmiljøinstitutt (STAMI) ved Toksikologisk ekspertgruppe for administrative normer (TEAN) bidrar med faglige vurderinger i dette arbeidet. TEAN vurderer og evaluerer de aktuelle SCOEL dokumentene, presiserer kritiske effekter og vurderer behov for korttidsverdier ut i fra den foreliggende dokumentasjonen. Videre søker og evaluerer TEAN nyere litteratur etter utgivelsen av dokumentet. TEAN bruker kriteriene gitt i SCOEL's metodedokument, "Methodology for the derivation of occupational exposure limits: Key documentation (version 7, June 2013)". Dette er inkludert i TEANs Metodedokument del B (Prosedyre for utarbeidelse av toksikologiske vurderinger for stoffer som skal implementeres i det norske regelverket for grenseverdier etter direktiv fra EU-kommisjonen) utarbeidet for denne revisjonen.

Informasjon om bruk og eksponering i Norge innhentes fra Produktregisteret, EXPO databasen ved STAMI og eventuelle tilgjengelige måledata fra virksomheter/næringer. Beslutningsprosessen skjer gjennom drøftingsmøter der Arbeidstilsynet, Næringslivets hovedorganisasjon/Norsk Industri og Landsorganisasjonen i Norge deltar, samt orienteringsmøter og offentlig høring. Konklusjonene fra høringen med forskriftsendringer og nye grenseverdier forelegges Arbeids- og sosialdepartementet som tar den endelige beslutningen.



# Innledning

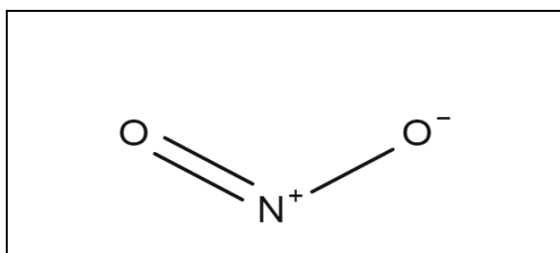
Dette grunnlagsdokumentet omhandler vurderingsgrunnlaget for fastsettelse av grenseverdi for nitrogendioksid. Innholdet bygger spesielt på anbefalinger fra Scientific Committee on Occupational Exposure Limits (SCOEL) i EU for nitrogendioksid (vedlegg 1), samt vurderinger og kommentarer fra Toksikologisk Ekspertgruppe for Administrative Normer (TEAN).

## 1. Stoffets identitet

Nitrogendioksid og dets molekylformel, synonymer av stoffets navn, stoffets identifikasjonsnummer i Chemical Abstract Service (CAS-nr.), European Inventory of Existing Commercial Chemical Substances (EINECS-nr. og/eller EC-nr.) og Indeks-nr. der disse er kjent er gitt i tabell 1. Strukturformel av stoffet er vist i figur 1.

**Tabell 1.** Stoffets navn og identitet.

<b>Navn</b>	<b>Nitrogendioksid</b>
<b>Molekylformel</b>	<b>NO<sub>2</sub></b>
<b>Synonymer</b>	Nitrogenoksid, nitrogenperoksid
<b>CAS-nr.</b>	10102-44-0
<b>EC-nr.</b>	233-272-6
<b>Indeks-nr.</b>	007-002-00-0



**Figur 1.** Strukturformel av nitrogendioksid.

## 2. Fysikalske og kjemiske data

Nitrogendioksid er en giftig, rødbrun gass med stikkende lukt.

Det vises til tabell 2 for fysikalske og kjemiske data for nitrogendioksid.

**Tabell 2.** Fysikalske og kjemiske data for nitrogendioksid (NO<sub>2</sub>).

Molekylvekt (g/mol)	46,01
Fysisk tilstand (20 °C, 101.3 kPa):	Brun væske under kokepunkt
Smeltepunkt (°C):	- 9,4
Kokepunkt (°C)	21,2
Tetthet (25 °C) (g/ml)	2,62
Damp tetthet (luft = 1):	1,58
Damptrykk (20 °C) (kPa)	96, 52 (80 °C) <sup>1</sup>
Omregningsfaktor (20 °C, 101 kPa)	1 ppm = 1.91 mg/m <sup>3</sup> 1 mg/m <sup>3</sup> = 0.523 ppm

<sup>1</sup>Data hentet fra SCOEL/SUM/53, juni 2014

## 2.1. Forekomst og bruk

NO<sub>2</sub> oppstår blant annet ved forbrenningsreaksjoner av organiske stoffer. Produksjon av NO<sub>2</sub> som endelige produktet er begrenset, men NO<sub>2</sub> dannes i betydelige mengder som mellomprodukt i industrielle aktiviteter i hele EU, spesielt i salpetersyre og gjødsel produksjon.

NO<sub>2</sub> kan dannes ved mange prosesser som forekommer i arbeidsmiljøet, for eksempel ved sveising, ved sprengningsarbeid, fra bensinmotorer og særlig dieselmotorer, enkelte ganger ved legging av silo og ved bruk av salpetersyre i galvanotekniske prosesser. Ofte dannes NO i tillegg til NO<sub>2</sub> ("nitrose gasser"). NO oksideres i tilfelle gjerne videre til NO<sub>2</sub>.

## 3. Grenseverdier

### 3.1 Nåværende grenseverdi

Nåværende grenseverdi (8 timer) i Norge for nitrogendioksid er: 0,6 ppm, 1,1 mg/m<sup>3</sup> med fotnote 12, fastsatt i 2007.

*Fotnote 12: Enkelte bedrifter vil av teknisk-økonomiske årsaker ikke kunne overholde denne verdien. Det er disse bedriftenes ansvar å dokumentere et forsvarlig arbeidsmiljø. Det forutsettes at bedriften(e) har eller er tilsluttet bedriftshelsetjeneste, og at eksponerte arbeidstakere gjennomgår egnede helseundersøkelser.*

### 3.2 Grenseverdi fra EU

Den europeiske vitenskapskomiteen, SCOEL foreslår:

IOELV (Indicative Occupational Exposure Limit Value) (8 timer): 0,5 ppm, 0,96 mg/m<sup>3</sup>

STEL (Short Term Exposure Limit): 1 ppm, 1,91 mg/m<sup>3</sup> som korttidsverdi.



I direktiv 2017/164/EU er det gitt en utvidet implementeringsfrist for de foreslåtte grenseverdiene til karbonmonoksid, nitrogenmonoksid og nitrogendioksid for bransjene *gruvedrift under jord* og *tunnel virksomhet*. Innføring av de nye grenseverdiene skal skje senest 21. august 2023 og dagens grenseverdier er i mellomtiden gjeldende i disse bransjene.

### 3.3 Grenseverdier fra andre land og organisasjoner

Nåværende grenseverdier (8 timer) for nitrogendioksid fra andre land og organisasjoner er gitt i tabell 3 nedenfor.

**Tabell 3.** Grenseverdier for nitrogendioksid fra andre land og organisasjoner. Land og organisasjoner som ikke har grenseverdier eller korttidsverdier for nitrogendioksid er markert med -.

Land Organisasjon	Grenseverdi (8 timer)	Korttidsverdi (15 min)	Anmerkning Kommentar
Sverige <sup>1</sup> NO <sub>2</sub> Avgasser	2 ppm, 4 mg/m <sup>3</sup> 1 ppm, 2 mg/m <sup>3</sup>	5 ppm, 10 mg/m <sup>3</sup> -	1990 2005, Note 5
Danmark <sup>2</sup>	2 ppm, 4 mg/m <sup>3</sup>	-	2000, L L - En takverdi som ikke skal overskrides på noe tidspunkt
Finland <sup>3</sup>	1 ppm, 1,9 mg/m <sup>3</sup>	2 ppm; 3,8 mg/m <sup>3</sup>	2015
Storbritannia <sup>4</sup>	-	-	-
Nederland <sup>5</sup>	0,4 mg/m <sup>3</sup>	1 mg/m <sup>3</sup>	(2007)
ACGIH, USA <sup>6</sup>	0,2 ppm, 0,38 mg/m <sup>3</sup>	-	-
NIOSH, USA <sup>6</sup>	-	1 ppm, 1,8 mg/m <sup>3</sup>	-
OSHA, USA <sup>6</sup>	5 ppm, 9 mg/m <sup>3</sup>	-	C - Takverdi
Tyskland, MAK <sup>6</sup>	0,5 ppm, 0,95 mg/m <sup>3</sup>	I(1)	Gjelder korttidsverdi: I(1) - Overskridelsesfaktor D - Enten finnes det ingen data for en vurdering av skade på embryo eller foster eller så er tilgjengelige data ikke tilstrekkelig for klassifisering i en av gruppene A-C.
Tyskland, Myndighetene <sup>7</sup>	0,5 ppm, 0,95 mg/m <sup>3</sup>	-	5/2016 EU 2(I) - Overskridelsesfaktor

<sup>1</sup> Arbetsmiljöverkets Hygieniska gränsvärden AFS 2015:7,

<https://www.av.se/globalassets/filer/publikationer/foreskrifter/hygieniska-gransvarden-afs-2015-7.pdf>.

<sup>2</sup> At-vejledning, stoffer og materialer - C.0.1, 2007, <https://arbejdstilsynet.dk/da/regler/at-vejledninger/g/c-0-1-graensevaerdi-for-stoffer-og-mat>.

<sup>3</sup> Social og hälsövärdministeriet, HTP-värden, Koncentrationer som befunnits skadliga, Helsingfors, 2016, [http://julkaisut.valtioneuvosto.fi/bitstream/handle/10024/79110/STM\\_9\\_2016\\_HTTP-varden\\_2016\\_Ruotsi\\_22122016\\_NETTI.pdf](http://julkaisut.valtioneuvosto.fi/bitstream/handle/10024/79110/STM_9_2016_HTTP-varden_2016_Ruotsi_22122016_NETTI.pdf).

<sup>4</sup> EH40 andre utgave, 2013, <http://www.hse.gov.uk/pubns/priced/eh40.pdf>

<sup>5</sup> [http://www.ser.nl/en/oel\\_database.aspx](http://www.ser.nl/en/oel_database.aspx);

<http://www.ser.nl/en/grenswaarden/2%20butyne%201%204%20diol.aspx>

<sup>6</sup> Guide to occupational exposure values compiled by ACGIH, 2017.



<sup>7</sup> Baa, TRGS 900, oppdatert 2016, [https://www.baua.de/DE/Angebote/Rechtstexte-und-Technische-Regeln/Regelwerk/TRGS/pdf/TRGS-900.pdf;jsessionid=439FFF321DF2323E60F868CD08E9CD3A.s1t2?\\_blob=publicationFile&v=2](https://www.baua.de/DE/Angebote/Rechtstexte-und-Technische-Regeln/Regelwerk/TRGS/pdf/TRGS-900.pdf;jsessionid=439FFF321DF2323E60F868CD08E9CD3A.s1t2?_blob=publicationFile&v=2)

### 3.4 Stoffets klassifisering

Nitrogendioksid er i henhold til CLP Annex VI (Forordning EC No 1272/2008)<sup>1</sup>, tabell 3.1 (Liste over harmonisert klassifisering og merking av farlige kjemikalier). klassifisert og merket i ulike fareklasser, med faresetninger og koder, som gitt i tabell 4 nedenfor.

**Tabell 4.** Fareklasser, farekategori med forkortelse, merkekoder og faresetninger for nitrogendioksid<sup>1</sup>

Fareklasse Farekategori Forkortelse	Merkekode	Faresetning
Gasser under trykk <i>Press. Gas</i>		
Oksiderende gasser Kategori 1 <i>Ox. Gas 1</i>	H270	Kan forårsake eller forsterke brann; oksiderende
Etsende/irriterende for huden Kategori 1B <i>Skin Corr. 1B</i>	H314	Gir alvorlige etseskader på hud og øyne
Akutt giftighet Kategori 2 <i>Acute Tox. 2</i>	H330	Dødelig ved innånding

<sup>1</sup> CLP ((Forordning (EC) Nr. 1272/2008), <http://www.miljodirektoratet.no/Documents/publikasjoner/M259/M259.pdf>  
<https://echa.europa.eu/information-on-chemicals/cl-inventory-database>

### 3.5 Biologisk overvåking

For å vurdere grad av eksponering for forurensning i luften på arbeidsplassen kan man anvende konsentrasjonen av forurensningen i arbeidstakerens urin, blod eller utåndingsluft, eller annen respons på eksponeringen i kroppen. EU har satt verdier for dette kalt biologisk grenseverdi (BLV).

SCOEL fremmer ikke et forslag til biologisk grenseverdi for nitrogendioksid.

## 4 Toksikologiske data og helseeffekter

### 4.1. Kommentarer fra TEAN

SCOELs kriteriedokument er fra 2014, de skriver i teksten at siste litteratursøk ble gjennomført i desember 2013. Nyeste studie referert til i SCOEL dokumentet er fra 2010, i tillegg er det referert til publiserte retningslinjer, hvor den siste er fra 2012. Det er gjennomgått kriteriedokumenter utarbeidet





av MAK fra 2010. MAK har kommet frem til tilsvarende vurdering av 8-timers grenseverdi som SCOEL, men har vurdert at korttidseksponeringsverdier bør være lik 8-timers grenseverdi. Det er foretatt et søk i søkebasen PubMed på litteratur publisert etter 2013. Det er få nye studier som har undersøkt yrkeseksponering for nitrogendioksid i denne perioden. En studie fant effekter på lungefunksjon ved konsentrasjoner mellom 1.21-7.82 mg/m<sup>3</sup> (0.6-4 ppm) NO<sub>2</sub>. Det er ingen beskrivelse av hvilke andre forbindelser arbeiderne eksponeres for og gjennomsnittskonsentrasjonen angis å være 3.9 mg/m<sup>3</sup> (2 ppm). Etter TEAN sin mening gir ikke denne studien ny informasjon som vil endre SCOEL sine vurderinger.

SCOEL vurderer at det er pålitelige og tilstrekkelige humane data på NO<sub>2</sub> fra arbeidende populasjon til å sette en 8-timers grenseverdi basert ut i fra disse.

SCOEL anser den kritiske effekten til NO<sub>2</sub> å være irritasjon dypt i lungene noe som kan gi ødem og nedsatt gassdiffusjon.

NO<sub>2</sub> reagerer langsomt med vann og danner salpetersyre og salpetersyrling som har irritative til kaustiske effekter. Salpetersyre og -syrling kan også dannes i lungene og syrene og deres salter er funnet i blod og urin. I mennesker tas 80-90% av NO<sub>2</sub> opp via lungene og absorberes for det meste i nedre del av lungene.

Yrkeseksponering for NO<sub>2</sub> forekommer alltid i kombinasjon med andre komponenter.

Det er spesielt to store, relativt nye studier som trekkes frem i SCOEL-dokumentet. Den ene studien har undersøkt langtidseffekter hos steinkullgruvearbeidere. Her ble det ikke funnet en klar sammenheng mellom eksponering og negative effekter på lungefunksjon, studien inkluderte arbeidere som var høyest eksponert slik som dieselmaskin- og diesellokomotivførere. Arbeiderne ble eksponert for gjennomsnittlige konsentrasjoner på 0.21 og 0.52 ppm NO<sub>2</sub> (Morfeld et al 2010). Dette angir en NOAEC på 0.5 ppm i forhold til effekter på lungefunksjon. SCOEL mener andre studier underbygger denne verdien. Det er vist effekter ved 0.4 og 0.5 ppm for NO<sub>2</sub> hos saltgruvearbeidere (Lotz et al 2008), men denne studien klarte ikke å skille ut effekter forårsaket av andre komponenter. I en eldre studie hvor arbeidere i kullgruver ble undersøkt, fant de ingen effekter, konsentrasjonene varierte fra 0.02 til 0.08 ppm for NO<sub>2</sub> (Robertson et al 1984). Heller ikke i denne studien kunne påvirkningen av andre komponenter utelukkes.

I følge SCOEL bygger dyrestudier under en grenseverdi på 0.5 ppm.

For korttidseksponering skjer den kritiske effekten lokalt i lunger. Det er funnet en økning i bronkial reaktivitet etter eksponering for 1.5 ppm over 3 timer hos frivillige friske personer. Økt luftveismotstand er funnet etter eksponering for 2 ppm NO<sub>2</sub> i fire timer over fire dager. Korttidseksponering har vist forandringer i bronko-alveolær lavage væske etter 3-timers eksponering av frivillige for konsentrasjoner fra 1.5 ppm (Sandström et al 1992).

**Karsinogenitet:** Det er vist i dyrestudier at eksponering for NO<sub>2</sub> kan ha kreftfremkallende egenskaper i lunge hos rotte. Inhalerte nitrose gasser kan reagere med aminer i lunge og føre til dannelse av kreftfremkallende aminer (Druckrey og Preussmann 1962). Det er ikke funnet gode langtidsstudier for at NO<sub>2</sub> er kreftfremkallende for mennesker og SCOEL påpeker at det er behov for flere studier om dette skal vurderes.

**Sensitivitet:** Astmatikere synes å være spesielt sensitive for NO<sub>2</sub>-eksponering. Det er sett effekter på lungefunksjon hos astmatikere ved lavere konsentrasjoner enn hos friske eksponert for NO<sub>2</sub>.



TEAN har ikke funnet relevante nye data som bestrider vurderingene i SCOEL-dokumentet, hverken for 8-timers grenseverdi eller korttidsverdi.

## 5. Bruk og eksponering

Nitrogendioksid er en giftig, rødbrun gass med stikkende lukt. Nitrogendioksid oppstår blant annet ved forbrenningsreaksjoner av organiske stoffer. Produksjon av nitrogendioksid som endelige produktet er begrenset, men nitrogendioksid dannes i betydelige mengder som mellomprodukt i industrielle aktiviteter i hele EU, spesielt i salpetersyre og gjødsel produksjon.

### 5.1. Opplysning fra Produktregisteret

Data fra Produktregisteret er innhentet oktober 2016.

Nitrogendioksid inngår ikke i produkter som er registreringspliktige i Produktregisteret (2016). Det foreligger derfor ingen opplysninger om hvilke typer produkter stoffet inngår i og ingen opplysninger om bransjeanvendelse fra registeret.

### 5.2. Eksponering og måledokumentasjon

Nitrogendioksid kan dannes ved mange prosesser som forekommer i arbeidsmiljøet, for eksempel ved sveising, ved sprengningsarbeid, fra bensinmotorer og særlig dieselmotorer, enkelte ganger ved legging av silo og ved bruk av salpetersyre i galvanotekniske prosesser (uhell). Ofte dannes nitrogenmonoksid i tillegg til nitrogendioksid ("nitrose gasser"). Nitrogenmonoksid oksideres videre til NO<sub>2</sub>.

#### 5.2.1. EXPO- data

Rapporterte målinger av nitrogendioksid er hentet fra STAMIs eksponeringsdatabase EXPO.

Eksponeringsmålinger av nitrogendioksid registrert i EXPO er utført over flere år. STAMI har siden 1984 kartlagt arbeidsatmosfæren på flere arbeidsplasser der blant annet nitrogendioksid kan være et problem. Målinger av nitrogendioksid for perioden 1984-2015 viser i underkant av 600 prøver fra ulike bedrifter og næringer.

Frem til 2000 gjaldt hovedvekten av undersøkelsene arbeid med tunneler. Ved denne typen arbeid kan arbeidstakerne bli eksponert for nitrogendioksid både fra dieseldrevne anleggsmaskiner og kjøretøy, og fra sprengningsarbeid og nitrogendioksid -nivået kan derfor bli høyt.

Grenseverdien for nitrogendioksid ble revidert og ny grenseverdi fastsatt i 2007. Eksponeringsmålinger registrert i EXPO frem til 2007 ble da vurdert og var med å danne grunnlaget for den nye grenseverdien. Eksponeringsmålinger som fremstilles i denne vurderingen inneholder målinger registrert etter den nye grenseverdien ble innført.

Det er utført i alt 86 målinger av nitrogendioksid i perioden 2009 – 2015, hvorav 41 prøver er personbårne, 39 er stasjonære prøver og de resterende er ikke spesifisert.

Resultatene viser at kun to målinger er registrert over grenseverdien på 0,6 ppm (1,1 mg/m<sup>3</sup>). Begge



målingene var personbårne og målt under tankbilarbeid. Den aritmetiske middelerdi var henholdsvis 5,1 ppm målt i en 8 minutters periode og 0,63 ppm, målt over 5 minutter.

Det er registrert 12 personbårne målinger i næringskoden «Bygging av veier og motorveier» under arbeid i undersjøisk tunnel. Eksponeringsnivået lå mellom 0,04 – 0,26 mg/m<sup>3</sup>, med en gjennomsnittsverdi på 0,1 mg/m<sup>3</sup>. Det ble foretatt 7 målinger under produksjon av jordbruks- og skogbruksmaskiner av smeder under smiing. Luftkonsentrasjonene av nitrogen-dioksid var betydelig lavere enn grenseverdien, med en gjennomsnittsverdi på 0,08 mg/m<sup>3</sup>. I tillegg er det utført personbårne målinger på matroser under fortøyning av båt og billettsalg. Resultatene viste et gjennomsnitt på 0,22 mg/m<sup>3</sup>.

Arbeidstilsynet har etter pålegg mottatt målinger av nitrogen-dioksid i anleggsvirksomhet i perioden 2016 – 2017 hvor nitrogen-dioksid ble målt (over 0,5 – 1,5 timer) under ulike typer arbeid bak stoff. 16 målinger viste gjennomsnittsnivåer fra 0,16 – 1,17 ppm, med aritmetisk gjennomsnitt på 0,51 ppm, og toppverdi på 4,2 ppm. 4 prøver hadde toppverdier over 1 ppm, og flere topper lå opp mot denne verdien. Måledataene er begrensede, men representativ for målinger i lang tunnel før gjennomslag.

Av de 39 stasjonære prøvene var 18 registrert i næringskoden «Andre tjenester tilknyttet utvinning av råolje og naturgass» og målt blant annet i ulike verksteder, lager og blanderom. 15 prøver var målt over deteksjonsgrensen og viste målinger fra 0,04 – 0,34 mg/m<sup>3</sup>. 7 målinger i tunnel i forbindelse med produksjon av elektrisitet fra vannkraft viste et gjennomsnitt på 0,07 mg/m<sup>3</sup>.

#### 5.2.2. Prøvetakings- og analysemetode

I tabell 5 er anbefalte metoder for prøvetaking og analyser av nitrogen-dioksid presentert.

**Tabell 5.** Anbefalte metoder for prøvetaking og analyse av nitrogen-dioksid.

Prøvetakingsmetode	Analysemetode	Referanse
Direktevisende gass-sensor	Electrochemical sensor <sup>1</sup>	
Filter (5.0µm PVC)+cellulosefilter impregnert med NaI	Vis absorption spectrometry <sup>2</sup>	Metode utviklet av Stami, publiseres i 2018

<sup>1</sup> Elektrokjemisk sensor

<sup>2</sup> Synlig absorpsjons spektrofotometri

## 6. Vurdering

Toksikologiske data for nitrogen-dioksid (NO<sub>2</sub>) er beskrevet i SCOEL-dokumentet i vedlegg 1, og kommentert av STAMI (TEAN) i kapittel 4.

Den kritiske effekten ved nitrogen-dioksid eksponering er irritasjon dypt i lungene som kan gi ødem og nedsatt gassdiffusjon.

Ved yrkeseksponering forekommer nitrogen-dioksid og nitrogenmonoksid (NO) sammen og i tillegg ofte andre stoffer som støv og dieselpartikler. Dette gjør det utfordrende å bestemme en grenseverdi for nitrogen-dioksid da det er vanskelig å fastslå effekten forårsaket av de enkelte stoffene i

eksponeringen. SCOEL tok utgangspunkt i to store epidemiologiske studier for å kunne bestemme en grenseverdi. Studiene er langtidseksponering av gruvearbeidere i tyske saltgruver og steinkullgruver. SCOEL angir sin anbefaling til grenseverdi for nitrogendioksid på grunnlag av langtidseffekter hos steinkullgruvearbeidere (Morfeld et al 2010) hvor det ble fastsatt en NOAEC på 0,5 ppm for lungefunksjonseffekter. Dette er grunnlaget for deres anbefaling til grenseverdi på 0,5 ppm.

For korttidseksponering skjer den kritiske effekten lokalt i lunger. SCOEL baserer sin anbefaling til korttidsverdi på en studie der korttidseksponeringer av frivillige friske personer viste en økt bronkial reaktivitet ved eksponering for 1,5 ppm nitrogendioksid i 3 timer og økt luftveismotstand etter eksponering for 2 ppm i 4 timer (Sandström et al 1992). Ettersom studiet viser at en økt bronkial reaktivitet ved 1,5 ppm, foreslår SCOEL en korttidsverdi på 1 ppm.

Det er ikke funnet langtidstudier som viser at nitrogendioksid er kreftfremkallende for mennesker.

Hos personer med redusert luftveisfunksjon (astmatikere og personer med kronisk bronkitt) har nitrogendioksid effekter på luftveiene ved et lavere nivå enn hos personer med normale luftveier. Flere studier har sett på effektene av nitrogendioksid eksponering på lungefunksjon hos astmatikere og personer med kronisk lungesykdom. Det ble funnet at astmatikere har vist obstruktiv effekt på lungefunksjonen ved 0,3 til 0,5 ppm nitrogendioksid.

SCOEL foreslår ingen hudanmerkning, fordi hudopptak ikke ansees å bidra til opptak av nitrogendioksid.

Det foreligger et begrenset antall målinger (86) registrert i EXPO siden den gjeldende grenseverdien ble innført i 2007. Eksponeringsmålingene foretatt under tankbilarbeid viste verdier over grenseverdien, men det var kun foretatt 2 målinger som ikke gir nok data for å trekke konklusjoner om gjennomførbarheten av en lavere grenseverdi. De resterende målingene foretatt i de ulike næringer var alle registrert under grenseverdien som kan gi holdepunkter for at den anbefalte grenseverdien vil være mulig å overholde.

Målingene innhentet av Arbeidstilsynet beskrevet ovenfor i kapittel 5.2.1 fra tunneldriving, viser at så snart ventilasjonskapasiteten går ned, øker nivå av nitrogendioksid betraktelig, og langt over grenseverdi. Sammen med de øvrige målte verdier kan måleresultatene således tyde på at ny anbefalt grenseverdi kan være mulig å overholde med nødvendig tunnelventilasjon og andre tiltak. De målte toppverdiene kan tyde på at korttids grenseverdi også kan overholdes dersom det sørges for nok ventilasjon og ikke minst andre tiltak der det forurensende arbeid foregår. Enkelte høye verdier over grenseverdier kan imidlertid forekomme.

## 7. Konklusjon med forslag til ny grenseverdi

På bakgrunn av den foreliggende dokumentasjon og en avveining mellom de toksikologiske dataene og eksponeringsdata, foreslås at dagens grenseverdi for nitrogendioksid senkes og at fotnote 12 sløyfes. I tillegg foreslås en korttidsverdi for stoffet, anmerkning E (EU har fastsatt grenseverdi for stoffet) og at en fotnote tilføyes.

Forslag til ny grenseverdi, korttidsverdi, anmerkning og fotnote:

**Grenseverdi (8-timers TWA): 0,5 ppm, 0,96 mg/m<sup>3</sup>**

**Korttidsverdi (15 min): 1 ppm, 1,91 mg/m<sup>3</sup>**

**Anmerkning: S (Korttidsverdi) og E (EU har fastsatt grenseverdi for stoffet)**

### **Fotnote:**

Enkelte bedrifter vil av teknisk-økonomiske årsaker ikke kunne overholde grenseverdiene. Det er disse bedriftenes ansvar å dokumentere et forsvarlig arbeidsmiljø. Det forutsettes at bedriften(e) har en plan for reduksjon av eksponering og at man kan vise lavere verdier over tid. Arbeidstilsynet, ansattrepresentanter og verneombud skal konsulteres og informeres om årlige planer og oppnådde resultater.

## 8. Ny grenseverdi

Dette kapitlet utarbeides etter at ASD har fastsatt den nye grenseverdien.



## 9. Referanser

1. Carbone, U., Montuori, P., Novi, C., Triassi, M. Respiratory function in power plant workers exposed to nitrogen dioxide. *Occup Med (Lond)*. 2014 Dec;64(8):644-6.



**Vedlegg: SCOEL/SUM/53**





European  
Commission

# Recommendation from the Scientific Committee on Occupational Exposure Limits for Nitrogen Dioxide

*SCOEL/SUM/53  
June 2014*

Employment,  
Social Affairs  
and Inclusion





## Table of Contents

1. Substance identification, physico-chemical properties.....	3
2. Occurrence/use and occupational exposure .....	3
2.1. Analytical measurement.....	4
3. Health significance .....	4
3.1. Toxicokinetics .....	4
3.2. Acute toxicity.....	4
3.2.1. Human data .....	4
3.2.2. Animal data .....	6
3.3. Sensory threshold and irritation .....	6
3.4. Sensitisation.....	7
3.5. Repeated dose toxicity.....	7
3.5.1. Human data .....	7
3.5.2. Animal data .....	9
3.6. Genotoxicity .....	10
3.7. Carcinogenicity .....	11
3.8. Reproductive toxicity .....	11
3.8.1. Human data .....	11
3.8.2. Animal data .....	12
4. Recommendation .....	12
5. References.....	14

## Recommendation from the Scientific Committee on Occupational Exposure Limits for Nitrogen Dioxide

8-hour TWA:	0.5 ppm (0.955 mg/m <sup>3</sup> )
STEL (15-min):	1 ppm (1.91 mg/m <sup>3</sup> )
BLV:	-
Additional categorisation:	-
Notation:	-

This Recommendation is based on compilations by WHO (1997), DECOS (2004), US EPA (2008), DFG (2005 and 2010), ACGIH (2012) and National Research Council of the National Academies (2012). An additional literature search was performed in December 2013.

### 1. Substance identification, physico-chemical properties

Chemical name:	Nitrogen dioxide
Synonyms:	Nitrogen peroxide
Molecular formula:	NO <sub>2</sub>
EC No.:	233-272-6
CAS No.:	10102-44-0
Molecular weight:	46.01 g/mol
Boiling point:	21.2 °C
Melting point:	-9.3 °C
Vapour pressure (80 °C):	52 kPa
Conversion factors:	1.91 mg/m <sup>3</sup> = 1 ppm
(20 °C, 101.3 kPa)	1 mg/m <sup>3</sup> = 0.524 ppm

#### *EU harmonised classification:*

Press. Gas		
Ox. Gas 1	H270	May cause or intensify fire; oxidiser
Skin corr. 1B	H314	Causes severe skin burns and eye damage
Acute Tox. 2	H330	Fatal if inhaled

Depending upon the temperature, nitrogen dioxide is a colourless solid, yellow liquid or reddish-brown gas with an irritating odour. The vapour density is 1.58 times that of air. The odour threshold is 0.1–0.4 ppm (0.2–0.8 mg/m<sup>3</sup>) (US EPA 2008, DFG 2005).

### 2. Occurrence/use and occupational exposure

Nitrogen dioxide (NO<sub>2</sub>) is found in ambient air as a product of natural as well as human activities. Production of NO<sub>2</sub> as the final product is limited. Production as a chemical intermediate, particularly in nitric acid and fertiliser production is very large throughout the EU. Occupational exposure may occur in the chemical industry, during

gas welding, in agriculture (silos), in mining (explosives) and from exhaust from combustion engines in confined areas. It is also produced by various industrial emissions and in tobacco smoke. Combined exposure to NO<sub>2</sub> and nitrogen monoxide (NO) is ubiquitous. Oxidation of NO to NO<sub>2</sub> may easily occur in some circumstances (DFG 2005 and 2010).

### 2.1. Analytical measurement

There are different methods for the measurement of airborne NO<sub>2</sub> (electrochemical cell, chemoluminescence). The lowest detection limit (LDL) of 0.002 ppm is reached by using chemoluminescence. Chemoluminescence measurements are considered to represent the "gold standard" of NO/NO<sub>2</sub> analysis. However, there are restrictions to the applicability of this method in underground mining (especially coal mining) or at workplaces where heat is involved. By contrast, the "multiwarn" electrochemical device, which allowed personal air sampling, had an LDL of only 1 ppm, but was suitable for the measurement of even very short exposure peaks (1 min). This instrumentation was used in the German coal mine studies (Dahmann *et al* 2007 and 2009; see Section 3.5.1). However, since the publishing date of these studies technology has further advanced. The "multiwarn" is no longer manufactured, and there are now new multi-gas detectors and new electrochemical sensors commercially available. Application of new technology resulted in an electrochemical sensor with a lower detection limit of NO<sub>2</sub> of 0.04 ppm. This sensor can be used in gas detectors that are applied in coal mines in several countries including Europe, as they are ATEX approved for use in mining environments (communication by Dräger Safety AG & Co KGaA, Lübeck/Germany, to SCOEL, dated 18 October 2013).

## 3. Health significance

The critical effect of NO<sub>2</sub> is irritation of the deep compartment of the respiratory tract in both animals and man. NO<sub>2</sub> is well absorbed via the lungs. It is then likely to be incorporated into intermediary metabolism pathways and does not result in systemic effects.

### 3.1. Toxicokinetics

NO<sub>2</sub> reacts only slowly in water to form nitric acid and nitrous acid, both of which have irritative to caustic effects (Mücke and Wagner 1998). It was concluded from a study with monkeys with radioactively labelled NO<sub>2</sub> that nitric acid and nitrous acid are formed also in the respiratory tract. The acids or their salts were detected in blood and urine. Experimental studies also showed that NO<sub>2</sub> or its products can remain in the lungs for a longer period (WHO 1997). In man, 80–90 % of NO<sub>2</sub> is taken up via the respiratory tract during normal breathing, and over 90 % at maximum breathing. Dosimetric model calculations showed that NO<sub>2</sub> is absorbed mainly in the lower respiratory tract, where NO<sub>2</sub> accumulates particularly in the area between the conductive and respiratory airways (pulmonary acinus), in which morphological changes are observed (WHO 1997).

### 3.2. Acute toxicity

#### 3.2.1. Human data

In the majority of studies with healthy volunteers, no effects on lung function were observed after short-term exposure (a maximum of 4 hours) to NO<sub>2</sub> concentrations in the air of 0.1 to about 2 ppm (National Research Council of the National Academies 2012). Effects on lung function were described at even lower concentrations in two studies including subjects with asthma (Bylin *et al* 1985 and 1988, Kulle 1982).

Increased airway resistance was reported in several studies only at 2 ppm and above (DFG 2005). Evidence of increased bronchial reactivity was observed after NO<sub>2</sub> concentrations of 1.5 ppm and above (Frampton *et al* 1989 and 1991, Mohsenin 1987 and 1988).

In particular, a study by Frampton *et al* (2002) appears relevant and is therefore described in detail below.

In this study, 21 volunteers were exposed to NO<sub>2</sub> at 0, 0.6 and 1.5 ppm for 3 hours during physical exercise (10 minutes of physical exercise at a ventilation rate of 40 l/min alternating with 20 minutes without physical exercise) at one-week intervals. NO<sub>2</sub> gas (5 000 ppm) was mixed with pure air and introduced into an exposure chamber (45 m<sup>3</sup>) to generate the desired NO<sub>2</sub> concentration. More than 90 % of the desired NO<sub>2</sub> concentrations were reached within 4 minutes. This publication includes no data on the NO<sub>2</sub> analysis. It is assumed that the method described in the earlier publications by Frampton *et al* (1989 and 1991) was used. There, continuous measurement was applied using an EPA reference standard NO/NO<sub>x</sub> analyzer, which was calibrated with standardised NO<sub>2</sub> and NO gas and verified by means of a calorimetric method. The background concentrations of NO<sub>x</sub>, ozone, dust and SO<sub>x</sub> were recorded continuously, but the results were not reported. No significant changes of spirometric parameters were observed. No effect of *in vivo* exposure to NO<sub>2</sub> on the inactivation of influenza viruses *in vitro* in alveolar macrophages or in bronchial epithelial cells was found, as had been described in a previous study (Frampton *et al* 1989). After exposure, a slightly, but statistically significantly reduced haemoglobin concentration and a reduced haematocrit (no difference between men and women) resulting from a decreased erythrocyte count were found in the blood of the volunteers at both NO<sub>2</sub> concentrations. The authors pointed out that such effects had also been observed in another volunteer study at NO<sub>2</sub> concentrations of 1 and 2 ppm (Posin *et al* 1978). The leukocyte count (polymorphonuclear leukocytes, lymphocytes, monocytes and eosinophils) was also reduced in the blood, again at the low concentration and in both sexes in most cases. A differentiation of the blood lymphocyte populations revealed a lymphocyte CD4<sup>+</sup>/CD8<sup>+</sup> ratio, which was increased in men but decreased in women. Total protein and albumin concentrations were not altered in the bronchial and alveolar lavage. In the bronchial lavage, clearly increased number of polymorphonuclear leukocytes (as a marker of inflammation) were found in men and, to a lesser degree, in women after exposure to 1.5 ppm; lymphocyte count increases were observed in both sexes, but they were more pronounced at 0.6 ppm than at 1.5 ppm. As in the bronchial lavage, an increased number of CD4<sup>+</sup> cells and an increased ratio of CD4<sup>+</sup>/CD8<sup>+</sup> lymphocytes were found in the alveolar lavage particularly in men at both 0.6 and 1.5 ppm, whereas in women the number of CD4<sup>+</sup> cells was increased only at 1.5 ppm. The authors pointed out that other publications also described sex differences in the T lymphocyte concentrations. They concluded that NO<sub>2</sub> exposure reduced the number of circulating T lymphocytes. The authors also pointed out that all effects observed were slight and presumably of no clinical relevance to healthy persons (Frampton *et al* 2002).

The findings obtained in this study at 0.6 ppm appear not to be adverse, particularly as in other volunteer studies effects occurred only in ranges starting from 1.5 ppm. Thus, in the bronchoalveolar lavage fluid (BAL or BALF), initial signs of inflammatory reactions were observed at NO<sub>2</sub> concentrations of 1.5 and 2 ppm. These signs were, for example, a reduction of cytotoxic suppressor T cells and natural killer cells (1.5 ppm for 20 min every 2<sup>nd</sup> day, 6 times; Sandström *et al* 1992), an increased number of CD25<sup>+</sup> lymphocytes and HLA-DR<sup>+</sup> macrophages in the BALF and neutrophils in the bronchial fraction (2 ppm NO<sub>2</sub> for 4 hours on 4 consecutive days; Blomberg *et al* 1999), a lower fraction of CD4<sup>+</sup> cells and an increased number of neutrophils in the bronchial fraction (Solomon *et al* 2000), a reduced capacity for phagocytosis of

alveolar macrophages and a decrease in superoxide production as well as an increased number of polymorphonuclear leukocytes in the bronchial lavage (2 ppm NO<sub>2</sub> for 4 hours, once; Devlin 1999; see DFG 2005 and 2010 for details).

Effects of NO<sub>2</sub> on lung function and airway responsiveness in healthy individuals are reported to occur at concentrations above 1 ppm (1.9 mg/m<sup>3</sup>) (Folinsbee 1992, Bylin 1993). Exposure to 0.6 ppm (1.1 mg/m<sup>3</sup>) NO<sub>2</sub> for 3 hours reduced the efficiency of macrophages to inactivate influenza virus in 4 out of 9 healthy volunteers. Nevertheless, the observed differences in virus inactivation after air and NO<sub>2</sub> exposure did not reach statistical significance at the P < 0.05 level, as it was stated by the authors (Frampton *et al* 1989). An increase in blood glutathione content was reported following exposure of volunteers to 0.2 ppm (0.4 mg/m<sup>3</sup>) for 2 hours (Chaney *et al* 1981), but this effect is considered to be of less biological importance.

Studies on the effects of NO<sub>2</sub> on pulmonary function in asthmatics and in patients with chronic lung disease or bronchitis were reviewed *in extenso* by the US National Research Council of the National Academies (2012). Overall, these studies were found conflicting and inconclusive with regard to the derivation of limit values. It was concluded that some asthmatic subjects exposed to 0.3–0.5 ppm NO<sub>2</sub> may respond with either subjective symptoms or slight changes in pulmonary function, which were rated to be of no clinical significance. As a potential population subgroup with higher susceptibility to NO<sub>2</sub>, children (5–12 years old) were identified. As children of this age are not exposed occupationally, such studies are not considered here.

### 3.2.2. Animal data

For rats, the LC<sub>50</sub> after exposure for 15 minutes was given as about 200 ppm and after 60 minutes as 115 ppm (Carson *et al* 1962). The LC<sub>50</sub> values for various strains of rat after exposure to NO<sub>2</sub> for 16 hours were given as 39–56 ppm. Mice, depending on the strain, were found to have LC<sub>50</sub> values between 33 and 67 ppm after exposure for 16 hours, golden hamsters values of 22 ppm (females) and 28 ppm (males) and Hartley Guinea pigs values of 50 ppm (females) and 62 ppm (males) (Sagai and Ichinose 1987). For rabbits, the LC<sub>50</sub> after exposure for 15 minutes was found to be 315 ppm. For dogs, 53 ppm was given as the 50 % value of the LC<sub>50</sub> (corresponding to an LC<sub>50</sub> of about 105 ppm) after exposure to NO<sub>2</sub> for 60 minutes. In rats and rabbits, severely impaired breathing, eye irritation and reduced body weights were observed (Carson *et al* 1962).

Other animal experiments with single exposures are described in WHO (1997).

### 3.3. Sensory threshold and irritation

NO<sub>2</sub> has a penetrating odour. Depending on the study conditions, the perception threshold is between 0.1 and 0.2 ppm (Feldman 1974, Shalamberidze 1971). With slowly increasing concentrations the odour is not perceived until much higher concentrations have been reached (Henschler *et al* 1960), so the warning effect of the gas is poor under this condition.

The irritation threshold for NO<sub>2</sub> in air was given as 20–30 ppm (Henschler *et al* 1960). At the workplace, exposure to 25–75 ppm led to bronchitis or bronchopneumonia with complete restitution, 50–100 ppm to reversible bronchiolitis and focal pneumonitis, 150–200 ppm to lethal *bronchiolitis fibrosa obliterans* and more than 300 ppm to lethal pulmonary oedema and asphyxia (combined with methaemoglobinaemia) (Grayson 1956).

### 3.4. Sensitisation

There were no data pointing to a sensitisation potential.

### 3.5. Repeated dose toxicity

#### 3.5.1. Human data

In studies with occupational exposure to NO<sub>2</sub>, the exposure is always to a mixture of substances e.g. with Diesel motor emissions, NO, sulphur dioxide, smoke or mineral dust, which also impair the respiratory tract. Diesel motors are an important source of exposure to NO<sub>2</sub>. As a result of the exposure to a mixture of substances these studies are, however, unsuitable for the adequate evaluation of NO<sub>2</sub>-related effects.

In a questionnaire, 232 workers from 4 Diesel bus garages were asked about acute respiratory diseases. Lung function tests were carried out before and after the shift, and the NO<sub>2</sub> concentration was determined during the shift. Concentrations of other irritant gases were below the US national standards. The highest NO<sub>2</sub> concentrations in the air were  $0.56 \pm 0.38$  ppm, the lowest  $0.13 \pm 0.06$  ppm. Short-term personal air sampling often yielded exposure concentrations of over 1 ppm. The authors reported that the prevalence of acute respiratory symptoms was higher than expected only in the high exposure group ( $> 0.3$  ppm) (Gamble *et al* 1987).

In 259 workers of a salt mine who were exposed to Diesel motor emissions, chronic respiratory effects were reported in a questionnaire and X-rays and lung function tests were carried out. The NO<sub>2</sub> exposure was determined by personal air sampling and was on average between  $0.2 \pm 0.1$  ppm and  $2.5 \pm 1.3$  ppm. The authors reported that coughing was associated with age and smoking, and dyspnoea with age. There was no association, however, with the NO<sub>2</sub> exposure (Gamble *et al* 1983).

In a study with 20 000 coal mining workers exposed on average to NO<sub>2</sub> concentrations of 0.03 ppm and NO concentrations of 0.2 ppm, workers were not absent due to illness more frequently than unexposed persons because of airway infections. There were, however, problems in this study as regards the classification of exposure (Jacobsen *et al* 1988, cited in WHO 1997).

Robertson *et al* (1984) measured the levels of NO<sub>2</sub> and NO in nine British collieries between 1974 and 1979, which ranged, respectively, from 0.02 and 0.08 ppm and 0.11 and 1.23 ppm depending on colliery and type of work. No relationship was found between exposure and respiratory symptoms or decline in FEV<sub>1</sub>, nor was there any differences in symptoms or lung function between 44 pairs of men matched for age, dust exposure, smoking habit, coal rank and type of work but differing in respect of exposure to oxides of nitrogen.

In principle, reports on the effects of long-term occupational exposure relating to mixed exposures are difficult to assess with respect to deriving a recommendation for an occupation exposure limit (OEL), when a discrimination of the effects of nitrogen oxides from those of other exposure components (particulates, gases, basic variables) has not been made. In such studies, attention has been drawn to effects on lung functions and immune functions in miners (salt and iron) exposed to nitrogen oxides, together with other pollutants like Diesel exhausts, dust and CO, with NO<sub>2</sub> levels ranging from 0.14 to 0.5 ppm (Backé *et al* 2004, Adelothe *et al* 2006, Latza *et al* 2009).

Recently, two larger epidemiological studies have been performed in miners: by Lotz *et al* (2008) in German salt mines, and by Morfeld *et al* (2010) in German hard coal mines. The analytical part of the first was published by Dahmann *et al* (2007) and that

of the second study by Dahmann *et al* (2009). The first study (Lotz *et al* 2008) in salt mines was accompanied by a multitude of analytical measurements, covering 600–700 shift mean values in total. This data base allowed a very precise statistical analysis of the distribution of shift exposures: the 95<sup>th</sup> percentile of 8-hour shift data was slightly higher than the two-fold of the mean, both for NO and for NO<sub>2</sub>.

The generation of a comparable body of analytical data was not possible in the coal mine study (Morfeld *et al* 2010). The main reason for this deficit of the coal mine study was that the chemoluminescence instruments did not fulfil the very strict explosion safety requirements for underground coal mines. This led to relatively few analytical measurements reported in the coal mine study (e.g. 21 in Diesel locomotive drivers, 5 in blasting workers). However, on the basis of the pre-existing knowledge (Dahmann and Monz 2000, Dahmann *et al* 2007) it was anticipated that the general data distribution pattern of analytical data for NO/NO<sub>2</sub> in the coal miners would be similar to those found in the salt miners.

Lotz *et al* (2008) examined 410 and 463 miners (salt mines A and B) cross-sectionally; 75 and 64 % of the first cohort were again examined after a 5-year period. Exposure was measured by personal sampling. Personal lifetime exposure doses of salt dust, Diesel exhaust, NO<sub>2</sub> and NO were calculated for all miners. Dose-response relationships were calculated by multiple regression analysis. In the 5-year period, the adjusted (age, smoking etc.) effect of the exposure indicators resulted in a mean decrease of FEV<sub>1</sub> between -18 ml/year (mine A) and -10 ml/year (mine B). The personal concentrations related to this effect were 12.6 and 7.1 mg/m<sup>3</sup> inhalable dust, 2.4 and 0.8 mg/m<sup>3</sup> respirable dust, 0.09 and 0.09 mg/m<sup>3</sup> Diesel exhaust, 0.4 and 0.5 ppm NO<sub>2</sub>, and 1.7 and 1.4 ppm NO (mines A and B). Exposure was related to symptoms of chronic bronchitis only in mine B. The authors concluded that the effects found in both mines indicated that the prevailing mixed exposure may cause lung function impairment in salt miners exposed over a long period of time. In this study, it was not possible to determine the effects of a single exposure component (nitrogen oxides vs. dust, Diesel exhaust etc.) separately.

This limitation was avoided in the study of hard coal miners by Morfeld *et al* (2010) by use of General Estimation Equation (GEE) models. This allowed the discrimination of effects of nitrogen oxides from those of other exposure component variables. A longitudinal inception cohort study (1974–1998) was conducted on miners who started working underground at two coal mines between 1974 and 1979. The authors determined the number of shifts underground, the exposure to coal mine dust, quartz dust, NO, NO<sub>2</sub>, smoking behaviours, and the lung function parameters FVC, FEV<sub>1</sub> and FEV<sub>1</sub>/FVC. In total, 1 369 miners worked on average 3 017 shifts underground per person. The total mean respirable coal mine dust concentration was 1.89 mg/m<sup>3</sup> (quartz: 0.067 mg/m<sup>3</sup>), and nitrogen oxide concentrations were 0.58 ppm NO and 0.007 ppm NO<sub>2</sub>. However, the exposure in defined subgroups was consistently higher (mean 8-hour shift concentrations: Diesel engine drivers, 1.35 ppm NO and 0.21 ppm NO<sub>2</sub>; Diesel train drivers, 1.35 ppm NO and 0.52 ppm NO<sub>2</sub>; blasting specialists, 0.84 ppm NO and 0.014 ppm NO<sub>2</sub>; see Dahmann *et al* 2009). The GEE-regression models did not reveal clear adverse dust exposure effects. Nitrogen oxides combined (NO + NO<sub>2</sub>) showed small, statistically insignificant, effects on lung function, which were not considered as being adverse. It was concluded that nitrogen oxide exposures, including those in the subgroups, showed no adverse influence on lung function in this long-term longitudinal study.

Regarding the distribution of exposure data, the 95<sup>th</sup> percentiles of 8-hour shift data (Dahmann *et al* 2009, Morfeld *et al* 2010) were slightly higher than the two-fold of the means, which was also supported by the data of the preceding study in salt miners (Dahmann *et al* 2007, Lotz *et al* 2008).

### 3.5.2. Animal data

Numerous animal studies with repeated exposure are described in WHO (1997). Many animal studies have shown that long-term (continuous) exposure to NO<sub>2</sub> concentrations of 5 ppm in air can cause emphysema, which has also been observed in man (WHO 1997, DFG 2005).

Several studies were conducted, in which Wistar rats were continuously exposed (24 hours/day) to NO<sub>2</sub> at 0, 0.4, 1.2 and 4 ppm for 1, 2, 4, 8, 12 and 16 weeks (Ichinose and Sagai 1982) and to NO<sub>2</sub> at 0, 0.04, 0.4 and 4 ppm for 4, 9, 18 and 27 months (Kubota *et al* 1987, Sagai and Ichinose 1987, Sagai *et al* 1984). The NO<sub>2</sub> concentrations were measured continuously using the Monitor Labs 8440-L Nitrogen Oxide Analyzer (Ichinose and Sagai 1982), but the measurement accuracy is not known, particularly at the very low concentrations of 0.4 and 0.04 ppm (DFG 2005).

After up to 16 weeks of continuous exposure of 6 rats per group to NO<sub>2</sub>, no significant alterations were observed at 0.4 ppm. At 1.2 ppm and above, the level of sulphhydryl (SH) groups not associated with proteins was significantly increased in the lungs (as evidence of increased glutathione dependent enzyme activities); glutathione reductase activity was significantly increased in week 4 and ethane exhalation was slightly increased in the first 4 weeks. At 4 ppm, ethane exhalation was considerably increased in the first 4 weeks, the increase still being significant at the end of exposure. The following significant lung changes were also observed at this concentration: increased thiobarbituric acid reactants and significantly increased glutathione reductase, glucose-6-phosphate dehydrogenase and superoxide dismutase activities (Ichinose and Sagai 1982). When the NOAEC of 0.4 ppm is converted from continuous 24-hour exposure to 8-hour daily exposure, it would theoretically correspond to a concentration of about 1.2 ppm.

After up to 27 months of exposure of 3–6 rats per group, ethane exhalation was significantly increased at 0.04 ppm and above. From 0.4 ppm, the mean thickness of the air-blood barrier was increased slightly after 18 months and significantly after 27 months. Moreover, there was some interstitial oedema and slight bronchiolar and alveolar epithelium changes. In the lungs, the non-protein SH group and thiobarbituric acid reactant levels were significantly increased and glutathione peroxidase activity was reduced. Hypertrophy and hyperplasia of the bronchiolar epithelium, Clara cell hyperplasia, interstitial fibrosis and type I and type II cell hypertrophy were observed at 4 ppm. Clear increases in lipid peroxidation and alterations in protective enzyme activities were also found (Kubota *et al* 1987, Sagai and Ichinose 1987, Sagai *et al* 1984). Ethane exhalation is also used as a sensitive and non-invasive test to detect lipid peroxidation in humans (Kneepkens *et al* 1994); however, ethane exhalation is a marker of effects that are not of adverse nature. Therefore, 0.04 ppm could be assessed under these conditions as the NOAEC and 0.4 ppm as the LOAEC. When the NOAEC of 0.04 ppm and the LOAEC of 0.4 ppm are converted from 24-hour exposure to 8-hour daily exposure, these may correspond to concentrations of about 0.12 ppm and 1.2 ppm, respectively. However, because of the continuous exposure, this study is not considered to be a suitable basis for recommending an OEL.

After a 15-day exposure of mice to NO<sub>2</sub>, changes in the distribution of leukocyte subpopulations in the BALF were observed at 20 ppm, but not at 10 ppm (Wegmann *et al* 2002/2003).

Brown Norway rats were used as a sensitive model of allergic diseases to detect immunological effects of NO<sub>2</sub> exposure and to investigate the activity of alveolar macrophages and the production of cytokines. Males and females were exposed to NO<sub>2</sub> at 0.2, 0.5 and 2 ppm during mating. In the case of the pups, which were also exposed, there are no data on the exact exposure period or duration. At 0.5 and



2 ppm, changes in the BALF were observed in the 8- and 12-week-old pups – in most cases in the animals which were exposed directly. No alterations were observed at 0.2 ppm (Kumae and Arakawa 2006).

In an inhalation study, which was conducted according to modern standards (BASF 2006a), 18 male Wistar rats per group were exposed to NO<sub>2</sub> in whole-body exposure chambers on 5 days for 6 hours/day. The desired NO<sub>2</sub> concentrations, which were calculated by means of the gas supply and the pump rate, were 0, 0.5, 5 and 20 ppm. The analytical concentrations, which were calculated by means of an infrared (IR) spectrophotometer with NO<sub>2</sub> calibration, were 0, 0.1, 4.9 and 19.2 ppm. The authors discussed the high relative humidity of 50 % with possible condensation of water and NO<sub>2</sub> at the apparatus and problems with the correct regulation of the supply of air and NO<sub>2</sub> as possible causes of the low NO<sub>2</sub> concentration measured in the low concentration group. In addition to the usual examinations (mortality, body weight gain, feed consumption, haematology, clinical chemistry, organ weight determinations, gross-pathological and histopathological organ examinations), lipid peroxidation (malondialdehyde formation) and 8-hydroxy-2-deoxyguanosine formation were investigated in lung homogenates, the BALF was examined for cellular components, and enzyme activities and cell proliferation and apoptosis were measured in the bronchi, bronchioles and alveoli at the end of the 5-day exposure period. No alterations were observed in the low exposure group. Histopathological alterations in the lungs (bronchoalveolar hyperplasia, mononuclear cell infiltration and alveolar histiocytes) and in the trachea (diffuse hyperplasia of the tracheal epithelium) and cell proliferation in the large and medium bronchi, terminal bronchioles and in the alveoli occurred at 5 ppm and above. Significantly increased lung weights, alveolar oedema, increased apoptosis rates in the medium and large bronchi as well as increased cell numbers, an increased number of macrophages and polymorphonuclear neutrophils, increased total protein levels and an increased activity of  $\gamma$ -glutamyltransferase and lactate dehydrogenase in the BALF were additionally found at 20 ppm. The authors specified a NOAEC of 0.5 ppm (BASF 2006a), but the accuracy of this concentration is unclear and it rather seems to be the detection limit of the method used. The LOAEC was 5 ppm. Since no increased malondialdehyde concentration was measured even at the high NO<sub>2</sub> concentration of about 20 ppm, the method used (total lung homogenates) cannot be regarded as sensitive enough to detect lipid peroxidation. Nor was a formation of 8-hydroxy-2-deoxyguanosine in lung homogenates observed, which means that this measurement may not be sensitive enough either.

In the subsequent subchronic (13-week) inhalation study (BASF 2006b), groups of 15 male and 10 female Wistar rats were exposed to NO<sub>2</sub> in whole-body exposure chambers at 0, 0.1, 0.5 and 1 ppm on 5 days/week for 6 hours/day. The analytical concentrations of 0.008, 0.25, 0.82 and 2.15 ppm, which were measured by means of an IR spectrophotometer with NO<sub>2</sub> calibration, were higher than the desired concentrations by about a factor of 2 since calibration was based on an incorrect correlation equation. At the end of the 13-week exposure period, the BALF was examined again and cell proliferation and apoptosis were measured in the lungs in addition to the usual examinations (mortality, body weight gain, feed consumption, haematology, clinical chemistry, organ weight determinations, gross-pathological and histopathological organ examinations). However, no substance-induced effects were observed up to 2.15 ppm. Therefore, the authors specified a NOAEC of 2.15 ppm for this study. No major measurement inaccuracies are expected at this concentration.

### 3.6. Genotoxicity

NO<sub>2</sub> is mutagenic in bacteria (Biggart and Rinehart 1987) and clastogenic in mammalian cells *in vitro* (Görsdorf *et al* 1990, Tsuda *et al* 1981). *In vivo*, no induction

of chromosome aberrations was observed in leukocytes and spermatocytes of mice exposed to NO<sub>2</sub> (Gooch *et al* 1977). Dose-dependent increases in mutations and in chromosome aberrations were seen in lung cells from rats exposed to NO<sub>2</sub> at 8, 15, 21 and 28 ppm (15, 29, 40 and 53 mg/m<sup>3</sup>) (Isomura *et al* 1984), but because of the low survival of the lung cells (10–15 %), this study is not considered to be conclusive evidence of *in vivo* genotoxicity. The available data therefore give no indication of a systemic genotoxic effect of NO<sub>2</sub>, but further studies may be required to evaluate a possible local genotoxicity on epithelia of the airways.

### 3.7. Carcinogenicity

The possibility that inhaled nitrous gases could react with amines of the mucous membranes of the respiratory tract to form carcinogenic nitrosamines was noted early on (Druckrey and Preussmann 1962). Valid long-term studies of the carcinogenicity of NO<sub>2</sub> are, however, not available. Long-term studies with limited validity did not yield evidence of carcinogenic effects of NO<sub>2</sub>. Initiation promotion studies yielded evidence of tumour-promoting effects of NO<sub>2</sub> in the rat lung.

### 3.8. Reproductive toxicity

#### 3.8.1. Human data

Pregnant women (n = 51) who were exposed to nitrogen oxides in the air (average NO<sub>2</sub> concentrations of 0.023 mg/m<sup>3</sup> with peak levels up to 0.239 mg/m<sup>3</sup>), drinking water (nitrate concentrations in well water of up to 400 mg/l) and food, were examined during the birth. Methaemoglobin (MetHb) in the blood of the mother and in the umbilical cord blood was determined as the effect marker, and blood lipids and glutathione as markers for oxidative stress. In the newborn babies, the birth weight, the APGAR index (evaluating the most important body functions of the baby) and clinical diagnoses at birth were recorded. MetHb concentrations of a maximum of 2 % in maternal blood and of a maximum of 2.8 % in cord blood were regarded as normal. Around 55 % of the maternal blood samples were above 2 % MetHb and about 20 % above 5 %. Around 80 % of the cord blood samples were above 2 % MetHb and about 45 % above 5 %. In the case of increased MetHb concentrations in maternal or cord blood, the values for glutathione (total and reduced) were decreased and those for lipid peroxides increased. A strong association was found between increased lipid peroxides in cord blood and adverse birth outcome. With premature births in particular, the levels of MetHb in cord blood were increased (Tabacova *et al* 1998). The study suggests that increased concentrations of MetHb and lipid peroxides in cord blood can lead to premature births and other impairments at birth. A correlation of the findings with NO<sub>2</sub> concentrations in the air is, however, not possible from this study. As nitrite in particular (formed from nitrate) is responsible for the formation of MetHb, the oral uptake of nitrate and nitrite via water and food is regarded in the study of Tabacova *et al* (1998) as having a decisive influence.

In an earlier study, MetHb levels in blood were increased by 1 % after exposure of volunteers to NO<sub>2</sub> concentrations of 20 ppm for 2 hours (Henschler and Lüdtke 1963). A significant increase in the MetHb concentration is therefore not to be expected when NO<sub>2</sub> concentrations are below 1 ppm in air. The possibility of overestimating the MetHb in cord blood due to the presence of foetal haemoglobin (Lynch *et al* 1998) must also be considered.

The influence of air pollution with regard to birth defects was investigated in a study. Associations between individual defects and carbon monoxide and ozone were found;

no such associations were found with the other substances (including NO<sub>2</sub>; Ritz *et al* 2002).

### 3.8.2. Animal data

The reproductive toxicology of NO<sub>2</sub> has not been adequately investigated so far.

## 4. Recommendation

SCOEL considers that available human data on NO<sub>2</sub> obtained in working populations are reliable for recommending an 8-hour OEL, as these are seconded by both experimental human studies of shorter duration and experimental animal studies.

The primary target to be considered is the deep respiratory tract. The long-term longitudinal study in hard-coal miners by Morfeld *et al* (2010) did not reveal clear adverse effects on lung function, under the conditions given in this study (Section 3.5.1). This study included sub-collectives of higher exposures, namely Diesel engine drivers and Diesel train drivers, with average shift exposures of 0.21 and 0.52 ppm NO<sub>2</sub>, respectively (Dahmann *et al* 2009). Regarding the distribution of exposures, the 95<sup>th</sup> percentile of the 8-hour shift data was about two-fold of the mean. Considering these data, the conclusion of a human NOAEC of 0.5 ppm NO<sub>2</sub> regarding lung function, under conditions of chronic occupational exposure, appears to be safe. This NOAEC is not contradicted by earlier studies in salt miners (Lotz *et al* 2008) and in British collieries (Robertson *et al* 1984), which could not discriminate effects of other components (dust and chemicals). Moreover, short-term experimental human studies (see below), showing first effects at about 1.5 ppm NO<sub>2</sub>, are consistent with this human NOAEC. This argues in favour of a recommended OEL for NO<sub>2</sub> at 0.5 ppm, which is primarily derived from human data.

A proof of plausibility of this derivation is provided by the available data in experimental animals. In this respect, most older experimental studies can not be considered as a reliable assessment basis, because exposures in these studies were continuous. However, there are recent inhalation studies in rats, performed according to modern standards, with exposures 6 hours/day for 5 days (Section 3.5.2). In these studies, a LOAEC of 5 ppm was obtained after 5 days of exposure (BASF 2006a) and a NOAEC of 2.15 ppm after 13 weeks of exposure (BASF 2006b). In interpreting the experimental NOAEC of 2.15 ppm the following must be considered: (i) There were problems in the analytical measurement of the NO<sub>2</sub> concentrations; the actual experimental exposure levels might have been overestimated. (ii) Uncertainties must be considered with regard to the only subchronic experimental exposure period. On the one hand, from comparison of the results of the subacute (5 days) range finding study with those of the subchronic (90 days) study, it might be concluded that the length of exposure did not play a major role. However, inhalation studies with continuous exposure suggest that effects occur at lower concentrations with an increase in the duration of exposure (from 16 weeks to 27 months; Ichinose and Sagai 1982, Kubuta *et al* 1987, Sagai and Ichinose 1987). Given these prevailing uncertainties, the experimental animal studies are in full support of an OEL of 0.5 ppm for NO<sub>2</sub>.

*Therefore, SCOEL recommends an OEL (8-hour TWA) for NO<sub>2</sub> of 0.5 ppm.*

For limiting short-term exposures (STEL), it must be noted that the critical effects are local, on the respiratory system (Section 3.2.1). Studies with volunteers show that increased bronchial reactivity can be observed after exposure to NO<sub>2</sub> concentrations of 1.5 ppm and more for 3 hours (Frampton *et al* 1989), and increased airway resistance

after exposure to 2 ppm for 4 hours for 4 days (Blomberg *et al* 1999). Marked changes in the BALF, a relevant method for determining toxic effects of NO<sub>2</sub> in the alveolar fluid, were evident after NO<sub>2</sub> concentrations of 1.5 ppm and more (Sandström *et al* 1992) and 2 ppm and more (Blomberg *et al* 1999, Devlin *et al* 1999, Solomon *et al* 2000). After exposure to NO<sub>2</sub> concentrations of 0.6 ppm, evidence of inflammation was inconsistent (Frampton *et al* 1989, 1991 and 2002). Since changes in the BALF were observed in volunteers after a 3-hour exposure to NO<sub>2</sub> at 1.5 ppm and above, a STEL (15 min) of 1.0 ppm is proposed with a sufficient intrinsic margin of safety.

There is no indication of relevant skin absorption. Therefore, no "skin" notation is considered necessary. The available data give no indication of systemic genotoxic effects. However, as pointed out above, further studies should evaluate a possible local genotoxicity on airway epithelia.

There are no human studies on effects of low concentrations of NO<sub>2</sub> at workplaces in asthmatics. The studies in miners do not answer this question, as asthmatics are not among the workforce in underground mines. However, the absence of irritative and inflammatory effects at the recommended OEL suggests that this should also be protective in asthmatics.

With regard to analytical measurement (see Section 2.1), no major difficulties are foreseen, based on new technological development.

The present Recommendation was adopted by SCOEL on 11 June 2014.

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